



# Hetero-Diels–Alder synthesis of the spiroketal fragment of reveromycin A

Mariana El Sous and Mark A. Rizzacasa\*

*School of Chemistry, The University of Melbourne, Parkville, Victoria 3010, Australia*

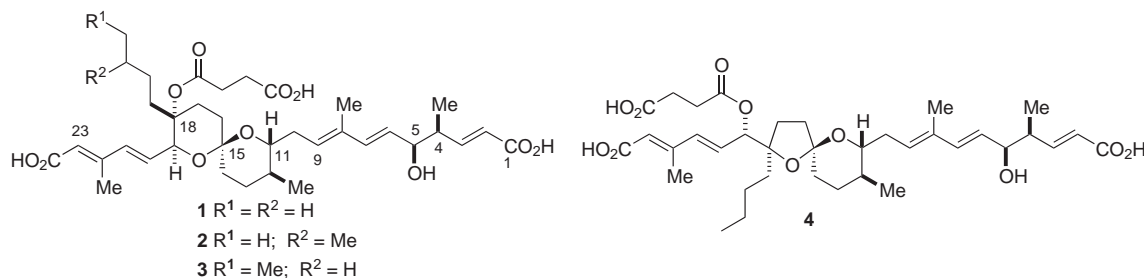
Received 15 April 2000; accepted 5 September 2000

## Abstract

The asymmetric synthesis of the 6,6-spiroketal fragment **15** of the epidermal growth factor inhibitor reveromycin A (**1**) is described. A hetero-Diels–Alder reaction was utilized to construct the 6,6-spiroketal **14** and subsequent stereoselective hydroboration provided reveromycin A spiroketal **15**. © 2000 All rights reserved. Elsevier Science Ltd.

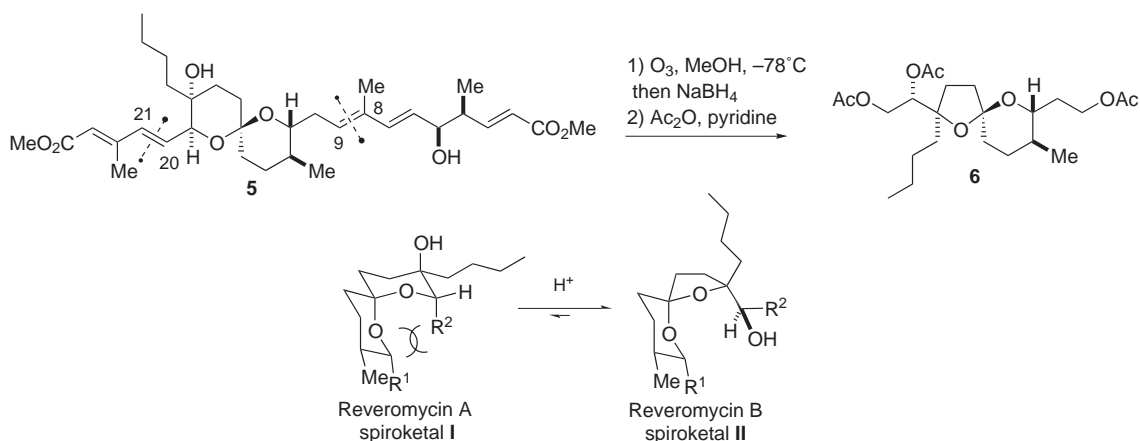
The reveromycins A (**1**), C (**2**), D (**3**) and B (**4**) are recent examples of natural products containing 6,6- and 5,6-spiroketal moieties which were isolated from a soil actinomycete belonging to the *Streptomyces* genus.<sup>1–3</sup> All the reveromycins act as inhibitors of the mitogenic activity of epidermal growth factor (EGF) which has been identified as a possible target for a new type of antitumor drug.

Reveromycin A (**1**) also exhibits antiproliferative activity against human tumor cell lines KB and K562 as well as antifungal activity.<sup>4</sup> The first total synthesis of (–)-reveromycin B (**4**) was reported in 1999<sup>5</sup> and a second total synthesis was communicated soon afterwards.<sup>6</sup> We have also recently reported the asymmetric total synthesis of (–)-reveromycin B (**4**)<sup>7</sup> in which the 5,6-spiroketal segment was synthesized by a hetero-Diels–Alder reaction followed by a novel stereoselective oxidation–rearrangement strategy.<sup>8</sup>



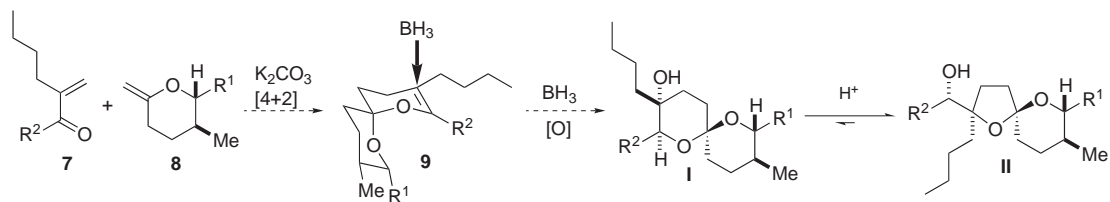
\* Corresponding author.

Reveromycin A (**1**) represents a greater synthetic challenge and so far, two independent groups have reported the synthesis of the 6,6-spiroketal fragment found in this compound.<sup>9,10</sup> In both approaches, the 6,6-spiroketal was formed by acid catalysis under *thermodynamic control* which lead to the generation of a mixture of bicyclic acetal/6,6-spiroketal<sup>9</sup> or 6,6-spiroketal isomers<sup>10</sup> due to the strain present in the reveromycin A type spiroketal. It has been shown that the spiroketal system in **1** is easily isomerized into the 5,6-spiroketal fragment found in **4**.<sup>9,10</sup> For example, reductive ozonolysis of desucinnoylated reveromycin A dimethyl ester **5**, followed by acetylation gave the 5,6-spiroketal **6** (reveromycin B type) exclusively (Scheme 1).<sup>9</sup> This result can be explained by the presence of a steric interaction in the 6,6-spiroketal system **I** which is alleviated upon acid induced rearrangement to the 5,6-isomer **II**.



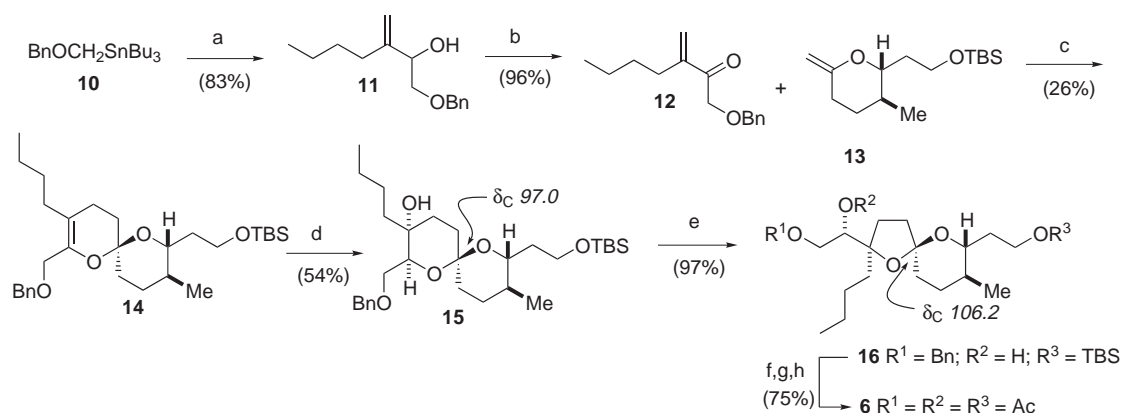
Scheme 1.

We elected to examine an alternative *kinetically controlled* approach to the 6,6-spiroketal system of **1** which utilizes a hetero-Diels–Alder reaction to construct the spiroketal system in a stereoselective manner (Scheme 2).<sup>11</sup> A [4+2] cycloaddition between a heterodiene **7** and a methylene pyran **8** should provide the 6,6-spiroketal **9** as one isomer where the stereochemistry at the spiro centre is controlled in the Diels–Alder reaction by the anomeric effect.<sup>8,12</sup> Stereo- and regioselective hydroboration of the alkene **9** should then occur from the less hindered face as shown to give the reveromycin A type 6,6-spiroketal **I** exclusively. The 6,6-spiroketal **I** should rearrange to the 5,6-isomer **II** upon treatment with a catalytic amount of acid.



Scheme 2.

To test the above proposal, a cycloaddition between the appropriately functionalized diene **12** and the optically pure methylene pyran **13**<sup>8</sup> (available in nine steps from 1,3-propanediol) was investigated (Scheme 3). Lithiation of the stannane **10**<sup>13</sup> and addition of the resultant anion to butylacrolein gave the alcohol **11**.<sup>14</sup> Oxidation with Dess–Martin reagent then provided the sensitive diene **12** which was immediately allowed to react with dienophile **13** in the presence of  $K_2CO_3$ <sup>8</sup> at 110°C to give the desired spiroketal **14** as the only detectable isomer. Other products from this reaction included *endo*-isomerized pyran as well as unidentified byproducts from the base induced degradation of diene **12**. Spiroketal **14** was then subjected to hydroboration and oxidation to provide the reveromycin A spiroketal **15**,<sup>15</sup> again as only one diastereoisomer, which exhibited a chemical shift for the spiro carbon (97.0 ppm) that is characteristic for the 6,6-spiroketal.<sup>8–10</sup> As expected, compound **15** completely rearranged to the reveromycin B 5,6-spiroketal **16** upon exposure to a catalytic amount of camphorsulfonic acid in  $CH_2Cl_2$ . The spiro carbon atom in 5,6-spiroketal **16** exhibited a signal at 106.2 ppm in the  $^{13}C$  NMR spectrum<sup>8,10</sup> and the structure was further confirmed by its conversion into the known triacetate **6**<sup>16</sup> which was identical to the naturally derived<sup>9</sup> and synthetic compounds<sup>5,9,10</sup> by all the usual criteria.



Scheme 3. Reagents and conditions: (a)  $n\text{BuLi}$ , THF,  $-78^\circ\text{C}$ , then butylacrolein; (b) Dess–Martin reagent,  $CH_2Cl_2$ , rt, 2.5 h; (c)  $K_2CO_3$ , 110°C, 22 h; (d)  $BH_3 \cdot THF$ ,  $0^\circ\text{C}$ , 2.5 h then  $NaOH/H_2O_2$ , rt, 2 h (e) cat. CSA,  $CH_2Cl_2$ , 2 h, rt; (f) TBAF, THF, 16 h, rt; (g)  $Pd(OH)_2$ ,  $H_2$ , MeOH, 2 h, rt; (h)  $Ac_2O$ , pyridine, DMAP, rt, 16 h

In conclusion, we have achieved an asymmetric synthesis of the reveromycin A 6,6-spiroketal via a hetero-Diels–Alder strategy. The synthesis is short (11 steps from 1,3-propanediol), convergent and leads to the production of the desired 6,6-spiroketal isomer **15** exclusively. In addition, the methodology could easily be applied to the synthesis of the other reveromycins C (**2**) and D (**3**). Studies towards the total synthesis of this family of compounds are underway.<sup>17</sup>

## Acknowledgements

We gratefully acknowledge the Australian Research Council for financial support.

## References

1. Takahashi, H.; Osada, H.; Koshino, H.; Kudo, T.; Amano, S.; Shimizu, S.; Yoshihama, M.; Isono, K. *J. Antibiot.* **1992**, *45*, 1409–1413.
2. Koshino, H.; Takahashi, H.; Osada, H.; Kiyoshi, I. *J. Antibiot.* **1992**, *45*, 1420–1427.
3. Ubukata, M.; Koshino, H.; Osada, H.; Isono, K. *J. Chem. Soc., Chem. Commun.* **1994**, 1877–1878.
4. Takahashi, T.; Osada, H.; Koshino, H.; Sasaki, M.; Onose, R.; Nakakoshi, M.; Yoshihama, M.; Isono, K. *J. Antibiot.* **1992**, *45*, 1414–1419.
5. Drouet, K. E.; Theodorakis, E. A. *J. Am. Chem. Soc.* **1999**, *121*, 456–457.
6. Masuda, T.; Osako, K.; Shimizu, T.; Nakata, T. *Org. Lett.* **1999**, *1*, 941–944.
7. Cuzzupe, A. N.; Hutton, C. A.; Lilly, M. J.; Mann, R. K.; Rizzacasa, M. A.; Zammit, S. C. *Org. Lett.* **2000**, *2*, 191–194.
8. McRae, K.; Rizzacasa, M. *J. Org. Chem.* **1997**, *62*, 1196–1197.
9. Shimizu, T.; Kobayashi, R.; Osako, K.; Osada, H.; Nakata, T. *Tetrahedron Lett.* **1996**, *37*, 6755–6758.
10. Drouet, K. E.; Ling, T.; Tran, H. V.; Theodorakis, E. A. *Org. Lett.* **2000**, *2*, 207–210.
11. (a) Paul, R.; Tchelitcheff, S. *Bull. Soc. Chim. Fr.* **1954**, 672–678. (b) Ireland, R. E.; Häbich, D. *Chem. Ber.* **1981**, *114*, 1418–1427. (c) Sauv e, G.; Schwartz, D. A.; Ruest, L.; Deslongchamps, P. *Can. J. Chem.* **1984**, *62*, 2929–2935. (d) Tietze, L. F.; Schneider, C. *J. Org. Chem.* **1991**, *54*, 2476–2481.
12. Deslongchamps, P. *Stereoelectronic Effects in Organic Chemistry*; Pergamon: Elmsford, NY, 1983.
13. Still, W. C. *J. Am. Chem. Soc.* **1978**, *100*, 1481–1487.
14. All new compounds were fully characterized by  $^1\text{H}$  and  $^{13}\text{C}$  NMR, IR,  $[\alpha]_{\text{D}}$  and HRMS or combustion analysis.
15. Data for spiroketal **15**: colorless oil (62 mg, 54%);  $R_{\text{f}}$  0.32 (10% EtOAc/petrol);  $[\alpha]_{\text{D}}^{21} +34.7$  ( $c$  0.75,  $\text{CH}_2\text{Cl}_2$ ); IR (film)  $\nu_{\text{max}}$  3446, 2957, 2861, 1650, 1459, 1253  $\text{cm}^{-1}$ ;  $^1\text{H}$  NMR (300 MHz,  $\text{CDCl}_3$ )  $\delta$  0.05 (s, 6H), 0.82 (d,  $J=6$  Hz, 3H), 0.88 (s, 9H), 0.91 (t,  $J=6.6$  Hz, 3H), 1.20–1.88 (m, 17H), 2.30 (m, 1H), 2.78 (s, 1H), 3.56–3.73 (m, 5H), 4.55 (ABq,  $J=11.7$  Hz, 2H), 7.26–7.34 (m, 5H);  $^{13}\text{C}$  NMR (75.5 MHz,  $\text{CDCl}_3$ )  $\delta$  -5.21, -5.23, 14.1, 17.9, 18.3, 23.4, 24.5, 26.0, 27.5, 29.7, 29.8, 31.8, 33.6, 34.8, 36.3, 59.2, 70.1, 71.3, 72.8, 73.6, 76.0, 97.0, 127.8, 127.9, 128.5, 137.5; HRMS (ESI) calc. for  $\text{C}_{30}\text{H}_{52}\text{O}_5\text{SiNa}$  [ $\text{M}+\text{Na}^+$ ]: 543.3482. Found: 543.3464.
16. Data for 5,6-spiroketal **6**: colorless solid, m.p. 58–59°C;  $R_{\text{f}}$  0.28 (15% EtOAc/petrol);  $[\alpha]_{\text{D}}^{29} +45.6$  ( $c$  1.0,  $\text{CHCl}_3$ ); lit.<sup>9</sup> (naturally derived)  $[\alpha]_{\text{D}}$  +39.1 ( $c$  0.13,  $\text{CHCl}_3$ ); lit.<sup>9</sup> (synthetic)  $[\alpha]_{\text{D}}$  +44.3 ( $c$  0.18,  $\text{CHCl}_3$ ); lit.<sup>10</sup> (synthetic)  $[\alpha]_{\text{D}}^{25} +37.5$  ( $c$  1.0,  $\text{CHCl}_3$ ); IR (film)  $\nu_{\text{max}}$  2958, 2934, 2874, 1744, 1459, 1438, 1371, 1243, 1187, 1128, 1047  $\text{cm}^{-1}$ ;  $^1\text{H}$  NMR (300 MHz,  $\text{CDCl}_3$ )  $\delta$  0.85 (d,  $J=6.3$  Hz, 3H), 0.90 (t,  $J=7.2$  Hz, 3H), 1.20–1.97 (m, 17H), 2.02 (s, 3H), 2.03 (s, 3H), 2.06 (s, 3H), 3.48 (dt,  $J=2.7, 8.4$  Hz, 1H), 4.12–4.23 (m, 2H), 4.31 (m, 1H), 4.50 (dd,  $J=2.3, 12$  Hz, 1H), 5.17 (dd,  $J=2.3, 8.7$  Hz, 1H);  $^{13}\text{C}$  NMR (75.5 MHz,  $\text{CDCl}_3$ )  $\delta$  14.1, 17.7, 20.9, 21.0, 21.1, 23.2, 25.5, 29.0, 31.4, 32.2, 34.3, 34.5, 34.6, 38.4, 61.7, 63.9, 73.6, 76.5, 86.2, 106.9, 170.3, 171.1. Anal. calc. for  $\text{C}_{23}\text{H}_{38}\text{O}_8$ : C, 62.42; H, 8.65. Found: C, 62.19; H, 8.36.
17. Note added in proof. The total synthesis of reveromycin A has been reported recently: Shimizu, T.; Masuda, T.; Hiramoto, K.; Nakata, T. *Org. Lett.* **2000**, *2*, 2153–2156.